Magnetoresistance in the Double Perovskite Sr₂CrMoO₆

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The double perovskite Sr_2CrMoO_6 with $Cr^{3+}-O-Mo^{5+}$ pairs is ferrimagnetic ($T_N = 450$ K), but there is considerable site disorder. The disorder appears to be responsible for the low value of magnetization. At 5 or 6 T, the negative magnetoresistance (MR) is <5% at all temperatures, unlike Sr_2FeMoO_6 which shows MR upto 27%. Tunneling magnetoresistance behavior of Sr_2CrMoO_6 is comparable to that of Sr_2FeMoO_6 , but the magnitude is considerably smaller in the former. © 2000 Academic Press

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INTRODUCTION

Following the surge of papers on the colossal magnetoresistance in rare earth manganates, substantial tunneling magnetoresistance (TMR) was reported at room temperature in the double perovskite Sr_2FeMoO_6 (1). Sr_2FeMoO_6 is a ferrimagnet with a transition temperature (T_N) of 415 K. Electronic structure calculations support the presence of spin-polarized charge carriers in the ground state of this solid. The resistance and magnetoresistance behavior of Sr_2FeMoO_6 is attributed to the dominant role played by the spin-dependent scattering at the grain boundary or the magnetic domain boundary. Apart from the A_2FeMoO_6 (A = Ca, Sr, or Ba) systems (1–4), magnetoresistance studies have been reported on the A_2FeReO_6 with A = Ca, Sr, Ba (5, 6). All of these compounds are ferrimagnetic with transition temperatures above room temperature.

In Sr₂FeMoO₆, the Fe³⁺ ions $(3d^5; t_{2g}^3 e_g^2; S = \frac{5}{2})$ contributes localized spin-up electrons while the itinerant spindown electrons are contributed by the Mo⁵⁺ $(4d^1; t_{2g}^1; S = \frac{1}{2})$. Sr₂FeReO₆ has an electronic structure similar to that of Sr₂FeMoO₆, but with a higher charge carrier density due to the larger number of *d* electron in the Re⁵⁺ ion $(5d^2)$. This would be expected to affect the magnetoresistance (MR) behavior of this compound, but the values of MR reported in A_2 FeReO₆ are low compared to those in A_2 FeMoO₆ (5, 6). Gracia-Landa *et al.* (7) have recently reported the results of neutron diffraction studies of Sr_2FeMoO_6 which indicate that the Fe atoms occupy the 4a position and the Mo atoms the 4b position in the ordered perovskite (Fm3m) structure. Surprisingly, there appears to be no magnetic contribution from 4b sites. Such an observation supports the presence of Fe²⁺-O-Mo⁶⁺ pairs in Sr₂FeMoO₆. The mechanism for exchange interaction would, however, be different for the Fe³⁺-O-Mo⁵⁺ pairs compared to the Fe²⁺-O-Mo⁶⁺ pairs. We have investigated the magnetotransport properties of the double perovskite Sr_2CrMoO_6 in comparison with those of Sr_2FeMoO_6 . The major difference between these two oxides is that in the Cr compound, there can be no valence compensation between the Cr and the Mo ions, since Cr can only be in the $3 + \text{ state } (t_{2g}^3)$. This requires Mo to be in the 5 + state inSr₂CrMoO₆, thereby limiting the system only to the Cr³⁺-O-Mo⁵⁺ pairs. Sr₂CrMoO₆ is also reported to exhibit a magnetic transition around 473 K (8).

EXPERIMENTAL

Polycrystalline samples of Sr₂CrMoO₆ were prepared by two methods. Sr_2CrMoO_6 (I) was prepared by heating a thoroughly ground mixture of SrCO₃, MoO₃, and Cr₂O₃ at 1173 K in air for 3 h, following Kobayashi et al. (1) in the case of Sr₂FeMoO₆. The product so obtained was ground, pelletized, and heated at 1473 K for 2 h in a flowing H_2/Ar (14:86 cm³/mm) atmosphere. Preparations of both Sr_2Cr -MoO₆ and Sr₂FeMoO₆ could be obtained by this procedure. A sample of Sr₂CrMoO₆ was made under similar conditions by annealing at 1473 K for 6 h in a flowing H_2/Ar (14:86 cm³/mm) atmosphere. In many of the Sr₂CrMoO₆ (I) samples the particle size was relatively small. Some of the samples prepared by the procedure of Kobayashi et al. (1) had SrMoO₄ impurity. Sr₂CrMoO₆ (II) was prepared by a two-step procedure. A precursor oxide of the composition $Sr_2CrO_{3,5}$ was prepared by heating $SrCO_3$ and Cr₂O₃ at 1273 K in air. The resultant oxide was mixed with stoichiometric quantities of MoO₃ and MoO₂



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to obtain Sr_2CrMoO_6 (II). Pellets were heated at 1273 K in an evacuated sealed silica tube for 7 days with intermediate grinding to obtain a pure phase.

Powder X-ray diffraction (XRD) data were collected with a Siemens 5005 X-ray powder diffractometer. Patterns of Sr_2CrMoO_6 with and without $SrMoO_4$ are given in Fig. 1. Rietveld refinement on the data set was carried out using the GSAS software suite (9). Four-probe dc electrical resistivity as well as magnetoresistance measurements were carried out using a cryo-cooled superconducting magnet. Magnetic susceptibility measurements were carried out with a Lewis Coil Force Magnetometer. Magnetization measurements were carried out using a vibrating sample magnetometer (Lakeshore VSM 7300).

RESULTS AND DISCUSSION

The X-ray diffraction patterns of the polycrystalline samples of Sr_2CrMoO_6 and Sr_2FeMoO_6 are shown in Fig. 1. The diffraction patterns could be indexed on a cubic *Fm3m*



lattice of the double perovskite structure. We have carried out Rietveld refinement of the powder X-ray data of several of the Sr₂CrMoO₆ samples along with Sr₂FeMoO₆. The calculated profiles gave reasonably good fits as can be seen from Fig. 1. The atomic coordinates and *R* values for Sr₂CrMoO₆ are compared with those of Sr₂FeMoO₆ in Table 1. The *R* values as well as the goodness of fit $(\chi^2 \ge 0.95)$ are quite satisfactory. From the Rietveld refinement of the X-ray data, the percent order in the Sr₂CrMoO₆ (I) and Sr₂FeMoO₆ was found to be 65 and 80, respectively. In an early report in the literature (8), 50% ordering was reported in Sr₂CrMoO₆.

In Fig. 2a, we show the electrical resistivity data of the Sr_2CrMoO_6 samples I and II, prepared by the two methods. The resistivity of Sr_2CrMoO_6 (I) decreased on annealing for a longer duration (6 h), possibly due to oxygen loss creating $Cr^{3+}-O-Mo^{4+}$ units. We have therefore given the results of the sample of Sr_2CrMoO_6 (I) annealed for 2 h along with those of the Sr_2CrMoO_6 (II) prepared by the two-step procedure. The resistivity of sample I is an order of magnitude lower than that of sample II.

In Fig. 3, we show the dc magnetic susceptibility of a Sr₂CrMoO₆ (I) sample. In the inset we show the inverse susceptibility behavior. The compound is ferrimagnetic, with a transition temperature of ~450 K. The effective magnetic moment (μ_{eff}), calculated from the high-temperature data in the paramagnetic phase, is 2.94 BM. This value is low compared to the expected μ_{eff} of 4.24 BM for Cr³⁺ and Mo⁵⁺. The low μ_{eff} value can be explained if we assume the presence of Cr³⁺–O–Mo⁵⁺ pairs in the paramagnetic state with a ferrimagnetic type of interaction between the

 TABLE 1

 Atomic Coordinates of Sr₂CrMoO₆ and Sr₂FeMoO₆ Samples

	Atom	Site	x	у	Ζ	Frac
Sr ₂ CrMoO ₆ ^a	Sr	8 <i>c</i>	$\frac{1}{4}$	$\frac{1}{4}$	$\frac{1}{4}$	1.00
	Cr	4a	0	0	Ó	0.65(3)
	Мо	4a	0	0	0	0.35(3)
	Cr	4b	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	0.35(3)
	Мо	4b	$\frac{1}{2}$	$\frac{\tilde{1}}{2}$	$\frac{1}{2}$	0.65(3)
	0	24e	0.253(4)	ō	Ō	1.00
Sr ₂ FeMoO ₆ ^b	Sr	8 <i>c</i>	$\frac{1}{4}$	$\frac{1}{4}$	$\frac{1}{4}$	1.00
	Fe	4a	Ō	Ō	$\vec{0}$	0.80(4)
	Мо	4a	0	0	0	0.20(4)
	Fe	4b	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	0.20(4)
	Мо	4b	$\frac{\overline{1}}{2}$	$\frac{\overline{1}}{2}$	$\frac{\overline{1}}{2}$	0.80(4)
	Ο	24 <i>e</i>	0.251(4)	õ	õ	1.00

FIG. 1. X-ray diffraction patterns of (a,b) two samples of Sr_2CrMoO_6 (I), (c) Sr_2CrMoO_6 (II), and (d) Sr_2FeMoO_6 . The calculated profiles and the difference curves obtained from Rietveld analysis are shown. Asterisk indicates line due to $SrMnO_4$ impurity.

 ${}^{a}a = 7.840(3)$; w $R_{p} = 6\%$; $R_{p} = 4\%$. The fit was good for Sr₂CrMoO₆ (I) with or without SrMoO₄, and also for Sr₂CrMoO₆ (II). One could refine the structure on the basis of a monoclinic cell with additional data, but the *R* values were roughly the same.

 $^{b}a = 7.899(1); \text{ w}R_{p} = 4\%; R_{p} = 3\%.$



FIG. 2. (a) Temperature variation of resistivity of the Sr_2CrMoO_6 (I) sample annealed for 2 h. Effect of a magnetic field of 5 T is shown. Resistivity data of Sr_2CrMoO_6 (II) prepared by the two-step procedure are shown in the inset. (b) Plot of -%MR against temperature for Sr_2CrMoO_6 (I) annealed for 2 h.

two ions. A μ_{eff} of 2.94 BM would result if 94% of the ions form Cr³⁺-O-Mo⁵⁺ pairs. Annealing for a longer period (6 h) reduced the μ_{eff} possibly due to the presence of Cr³⁺-O-Mo⁴⁺ pairs along with Cr³⁺-O-Mo⁵⁺ pairs. From the magnetization measurements carried out with a vibrating sample magnetometer at a field of 1 T, we could obtain the magnetic moments in the ferrimagnetic phase of Sr₂CrMoO₆. The magnetic moment for the sample was 0.2 BM/(formula unit) at 45 K. This value is much lower than the expected value of 2.0 BM for the fully ordered phase of Sr₂CrMoO₆. On the other hand, the magnetic moment for Sr₂FeMoO₆ is ~1.9 BM at a field of 1 T at

300 K and reaches a value of ~2.9 BM at 4.2 K (1). Magnetization data of Sr_2CrMoO_6 (II) was no better than that of I. The low value of the magnetic moment in the case of Sr_2CrMoO_6 can arise from the absence of long-range magnetic ordering due to site disorder in the double perovskite.

The effect of magnetic fields on the electrical resistivity of the Sr_2CrMoO_6 samples prepared by the two methods is shown in Fig. 2a. We see that a magnetic field of 5 or 6 T causes only a small decrease in resistivity in the both of the samples. The magnetoresistance of Sr_2CrMoO_6 (I) at 5 T is plotted against temperature in Fig. 2b. The magnitude of MR at the lowest temperature is considerably smaller



FIG. 3. Temperature variation of the magnetic susceptibility of the Sr₂CrMoO₆ (I) sample.

(2-5%) than that of Sr₂FeMoO₆ (~27%). The presence of SrMoO₄ impurity in sample I does not appear to affect the magnitude of MR significantly just as in Sr₂FeMoO₆ (1, 7, 10). Garcia-Landa *et al.* (7) do not find the effect of SrMoO₄ impurity on MR in their sample of Sr₂FeMoO₆, even though the impurity was as much as 10%. We found little effect of the magnitude of SrMoO₄ impurity on the properties of Sr₂CrMoO₆ (I). The magnitude of magnetoresistance of Sr₂CrMoO₆ (II) without any SrMoO₄ is even smaller



FIG. 4. Variation of the resistivity with magnetic field in (a) Sr_2FeMoO_6 and (b) Sr_2CrMoO_6 (annealed for 2 h).

than in I, being less than 2%. Part of the reason for the low MR values in the Cr compound may be site disorder. Disorder is known to lower the MR in Sr₂FeMoO₆ as well (11).

In Fig. 4, we show the resistivity-magnetic field plots at 40 K for Sr_2FeMoO_6 and Sr_2CrMoO_6 (I) annealed for 2 h and exhibiting maximum MR. The ρ (H) curves are similar, showing that tunneling magnetoresistance (TMR) occurs in both of these materials. The resistivity of Sr_2FeMoO_6 changes from 0.185 Ω cm to 0.129 Ω cm in a magnetic field of 9 T (-%MR \approx 30). The resistivity of Sr_2CrMoO_6 , on the other hand, changes from 0.125 Ω cm to 0.117 Ω cm (-%MR \approx 6). The difference in the magnitude of MR may also be related to the greater site disorder in Sr_2CrMoO_6 .

In conclusion, the double perovskite Sr_2CrMoO_6 , where there can be no valence compensation between Cr and Mo ions, shows the effect of disorder on the magnetization and magnetoresistance, in that the values of both are rather small. The low value of the TMR in this compound is also likely to be related to disorder effects.

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